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(54) **COATINGS AND METHODS FOR
CORROSION DETECTION AND/OR
REDUCTION**

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106/400; 106/600; 106/789

(58) **Field of Classification Search** 427/8;
428/402.2; 523/205, 201

See application file for complete search history.

(56) **References Cited**

U.S. PATENT DOCUMENTS

4,273,671 A * 6/1981 Allinikov 252/301.19
6,022,501 A * 2/2000 Dexter et al. 264/4.7
6,075,072 A * 6/2000 Guilbert et al. 523/200
6,080,334 A * 6/2000 Heimann et al. 252/389.62
6,544,540 B2 4/2003 Van Koppenhagen
6,716,526 B2 * 4/2004 Weston et al. 428/402.2
7,192,993 B1 * 3/2007 Sarangapani et al. 523/200
2002/0081431 A1 * 6/2002 Schmidt 428/402
2003/0068824 A1 4/2003 Frankel et al.
2005/0176851 A1 * 8/2005 Cook 523/210

OTHER PUBLICATIONS

A. Kumar and L.D. Stephenson, Smart coatings, 23rd Army Science
Conference, Orlando, FL, Dec. 2002, www.asc2002.com/summaries/a/AP-16.pdf.

M. Kendig, "Past, Present and Future 'Smart' Protective Coatings,"
*Conference Advanced Research & Development of Coatings for Cor-
rosion Protection: Offshore Oil & Gas Operations Facilities, Marine
Pipeline & Ship Structures*, Apr. 14-16, 2004, Biloxi, MS, Organized
by Colorado School of Mines.

H. Yang and W. vanOoij, *Plasmas and Polymers*, 8(4): 297-323 Dec.
2003.

M. Kedig et al., *Progress In Organic Coatings*, "Smart Corrosion
Inhibiting Coatings," 47, 183 (2003).

* cited by examiner

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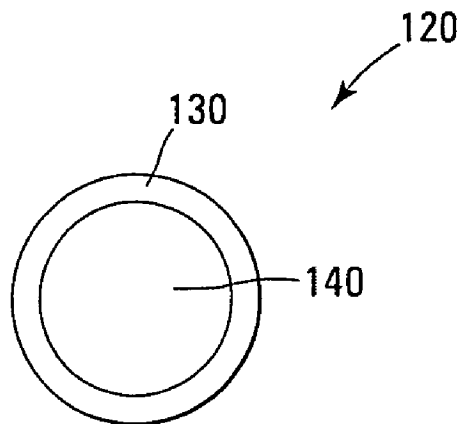
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(57) **ABSTRACT**

Coatings and methods are provided. An embodiment of the
coating includes microcapsules that contain at least one of a
corrosion inhibitor, a film-forming compound, and an indica-
tor. The microcapsules are dispersed in a coating vehicle. A
shell of each microcapsule breaks down in the presence of an
alkaline condition, resulting from corrosion.

17 Claims, 1 Drawing Sheet



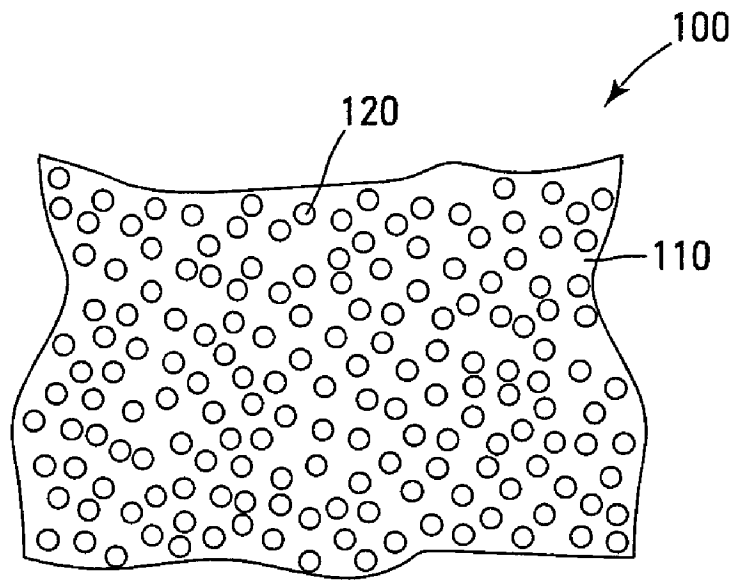


Fig. 1

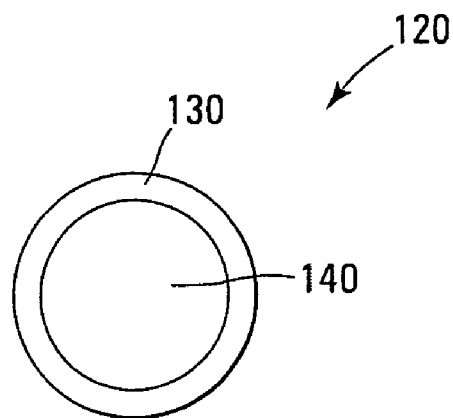


Fig. 2

1

COATINGS AND METHODS FOR CORROSION DETECTION AND/OR REDUCTION

STATEMENT OF GOVERNMENT INTEREST

The invention described herein was made in the performance of work under a NASA contract and by an employee of the United States Government and is subject to the provisions of Public Law 96-517 (35 U.S.C. §202) and may be manufactured and used by or for the Government for governmental purposes without the payment of any royalties thereon or therefore. In accordance with 35 U.S.C. §202, the contractor elected not to retain title."

TECHNICAL FIELD OF THE INVENTION

The present invention relates generally to corrosion and in particular the present invention relates to coatings and methods for corrosion detection and/or reduction.

BACKGROUND OF THE INVENTION

Metals and metal alloys are present in some shape or form in nearly every facet of our lives. Many metals and metal alloys are subject to corrosion that causes these metals and metal alloys to lose their structural integrity.

SUMMARY

The above-mentioned problems with metals and metal alloys and other problems are addressed by the present invention and will be understood by reading and studying the following specification.

For one embodiment, the invention provides a corrosion detection method, including applying a coating on a metal surface, the coating having microcapsules dispersed in a coating vehicle, the microcapsules containing an indicator, releasing the indicator in the presence of corrosion, and indicating the presence of the corrosion using the indicator, where releasing the indicator is caused by an alkaline condition, resulting from the corrosion, that breaks down shells of microcapsules exposed to the alkaline condition.

Further embodiments of the invention include methods and apparatus of varying scope.

BRIEF DESCRIPTION OF THE DRAWINGS

FIG. 1 illustrates an embodiment of a coating, according to an embodiment of the invention.

FIG. 2 illustrates an embodiment of a microcapsule, according to another embodiment of the invention.

DETAILED DESCRIPTION

In the following detailed description of the invention, reference is made to the accompanying drawings that form a part hereof, and in which is shown, by way of illustration, specific embodiments in which the invention may be practiced. In the drawings, like numerals describe substantially similar components throughout the several views. These embodiments are described in sufficient detail to enable those skilled in the art to practice the invention. Other embodiments may be utilized and structural, logical, and chemical changes may be made without departing from the scope of the present invention. The following detailed description is, therefore, not to be

2

taken in a limiting sense, and the scope of the present invention is defined only by the appended claims and equivalents thereof.

FIG. 1 illustrates a coating **100**, according to an embodiment. Coating **100** includes a coating vehicle **110**, such as clear or opaque paint, aqueous gel, water, etc., having microcapsules **120** dispersed within the coating vehicle **110**, e.g., by mixing etc. For embodiments including the aqueous gel, coating **100** is a temporary coating. For another embodiment, coating vehicle **110** may be a solvent, such as an aliphatic hydrocarbon, e.g., aliphatic petroleum distillates.

FIG. 2 is an enlarged view of a microcapsule **120**, according to another embodiment. Microcapsule **120** includes a shell **130** that contains an encapsulant **140**, such as a film-forming compound, an indicator, or a corrosion inhibitor, or various combinations thereof. A suitable film-forming compound may be a clear varnish, such as an acrylic varnish. A suitable indicator may be a pH indicator that changes color over the alkaline region, e.g., pHs from about 8 to about 10, such as phenolphthalein. Another suitable indicator is one that fluoresces, such as 7-hydroxycoumarin or coumarin, in the presence of or upon the oxidation of a metal or in the presence or upon the formation of a metal cation complex. A suitable corrosion inhibitor may be, sodium nitrate, camphor, polyamine fatty acid salts in a solvent, such as toluene, vegetable oil, or the like.

Shell **130** is broken down or disintegrates in the presence of a base (or an alkaline), e.g., having a pH above about 8, for releasing encapsulant **140**. Microcapsules having shells that break down under basic, or alkaline, conditions are known in the art. For example, a microcapsule **120** may be formed by forming an oil, e.g., toluene, vegetable oil, in water emulsion. A surfactant, such as xanthan gum, Attagel 40 (available from Engelhard Corporation, Iselin, N.J., USA), Petro BAF (available from Akzo Nobel Chemicals, Inc., Chicago, Ill., USA), and/or Reax 83 (available from MeadWestvaco Corporation, Stamford, Conn., USA) is added to the emulsion to evenly distribute the oil in the water. A pre-polymer, such as an amino resin pre-polymer, a cross-linking agent having one or more ester or thioester groups that are broken down under basic conditions, and encapsulant **140** are added to the oil phase. A catalyst, such as an inorganic acid, is added to the water phase. The oil in water emulsion is heated, causing a polymerization reaction to occur at the water-oil interface, which forms shell **130** around encapsulant **140**. It is the interaction of alkaline conditions with functional groups of the cross-linking agent that causes shell **130** to break down under alkaline conditions. Alternatively, encapsulant **140** can be released mechanically, such as by scratching or compressing a surface to which coating **100** is applied.

A basic, or alkaline, condition often forms in the presence of corrosion in a metal or a metal alloy, e.g., a basic condition often forms next to corrosion-induced pits, crevices, etc. For example, as is known in the art, when a drop of salt water is applied to steel, an anodic reaction occurs to produce a rust spot, and a cathodic reaction, e.g., involving water and oxygen, occurs to produce a basic condition. Therefore, when coating **100** is applied to a metal surface, and if there is corrosion, shells **130** of the microcapsules **120** exposed to the base condition will break down under the basic condition resulting from the corrosion, thereby releasing encapsulant **140**. Hereinafter, corrosion will refer to any chemical or electrochemical process involving the deterioration or degradation of metals, including pitting, crevice corrosion, or the like.

For one embodiment, coating **100** is temporarily applied to a metal-containing surface, e.g., by spraying, brushing, or rolling, to determine whether corrosion has occurred. For this

3

embodiment, coating vehicle **110** is an aqueous gel, and encapsulant **140** includes a suitable indicator, such as the pH indicator or the florescent indicator, as described above. In the presence of corrosion, the shells **130** of the microcapsules **120** break down, releasing the indicator. The pH indicator will change color or the florescent indicator will fluoresce to indicate the presence of corrosion. Coating vehicle **110** is not limited to an aqueous gel, but can be a clear paint or varnish, opaque paint or varnish, etc. that is substantially permanently applied to the metal-containing surface by spraying, brushing, or rolling. For another embodiment, coating **100** is applied substantially permanently to a bare metal surface as a primer coat that is clear coated. Another example of a temporary coating is a water suspension of microcapsules **120**. Examples of a metal-containing surface include a bare metal surface, a metal surface coated with, e.g., paint, varnish, epoxy, etc., a metal coated with a previous application of coating **100**, or the like. Other examples of a metal-containing surface include metal or metal alloy, e.g., a ferrous metal or alloy.

For some embodiments, e.g., where coating **100** is substantially permanently applied, encapsulant **140** may include the indicator or the corrosion inhibitor or both. If corrosion occurs, encapsulant **140**, due to shells **130** breaking down in the presence of the basic condition resulting from the corrosion, is released, and the corrosion inhibitor acts to prevent further corrosion, while the indicator indicates the corrosion location. For another embodiment, a trauma, such as a scratch, an impact, etc., to coating **100** may act to expose metal. The trauma can also break shells **130** so as to release corrosion inhibitor that acts to reduce the likelihood of the exposed metal of corroding.

For another embodiment, e.g., where coating **100** is substantially permanently applied, encapsulant **140** may include the film-forming compound, the film-forming compound and corrosion inhibitor, the film-forming compound, corrosion inhibitor, and indicator, or the film-forming compound and the indicator. For one embodiment, if coating **100** is exposed to trauma that causes shells **130** to break, the film-forming compound forms over at least a portion of the trauma and acts to reduce the likelihood of any exposed metal of corroding. For some embodiments, the corrosion inhibitor acts further to reduce the likelihood of corrosion of any exposed metal.

Note that if corrosion occurs at locations away from the trauma location, e.g., due to small breaks in the coating, such as chips, or adjacent the trauma location, corrosion inhibitor and film-forming compound will be released due to shells **130** breaking down in the presence of the basic condition, resulting from the corrosion. In other embodiments, the indicator will indicate the presence of the corrosion.

For one embodiment, a portion of microcapsules **120** in coating **100** contains corrosion inhibitor and another portion of microcapsules **120** in coating **100** contains indicator. For another embodiment, a portion of microcapsules **120** in coating **100** contains corrosion inhibitor, another portion of microcapsules **120** in coating **100** contains indicator, and yet another portion of microcapsules **120** in coating **100** contains film-forming compound. For another embodiment, a portion of microcapsules **120** in coating **100** contains film-forming compound and another portion of microcapsules **120** in coating **100** contains indicator. For some embodiments, microcapsules **120** having different contents are randomly distributed within coating vehicle **110** so that microcapsules **120** having the different functions of indicating, inhibiting, and/or

4

film-forming may be adjacent each other, as well as microcapsules **120** having like functions being adjacent each other.

CONCLUSION

Although specific embodiments have been illustrated and described herein, it will be appreciated by those of ordinary skill in the art that any arrangement that is calculated to achieve the same purpose may be substituted for the specific embodiments shown. Many adaptations of the invention will be apparent to those of ordinary skill in the art. Accordingly, this application is intended to cover any adaptations or variations of the invention. It is manifestly intended that this invention be limited only by the following claims and equivalents thereof.

What is claimed is:

1. A corrosion detection method, comprising:

applying a coating on a metal-containing surface, the coating comprising microcapsules dispersed in a coating vehicle, the microcapsules comprising shells encapsulating an indicator;

breaking down the shells by interaction of the shells and an alkaline condition naturally occurring in the presence of corrosion; and

releasing the indicator.

2. The detection method of claim 1, wherein indicating the presence of corrosion using the indicator comprises the indicator changing color in response to the alkaline condition.

3. The detection method of claim 1, wherein indicating the presence of corrosion using the indicator comprises the indicator fluorescing in response to the alkaline condition.

4. The detection method of claim 1, wherein the coating vehicle is selected from the group consisting of clear paint, clear varnish, opaque varnish, opaque paint, water, solvents, and aqueous gel.

5. A method of treating of a surface, comprising:

applying a coating to the surface, wherein the coating comprises microcapsule shells dispersed in a coating vehicle, the microcapsule shells containing a corrosion inhibitor;

breaking down the microcapsule shells through interaction of the shells and an alkaline condition naturally occurring in the presence of corrosion; and

releasing the corrosion inhibitor.

6. The method of claim 5, wherein the microcapsules further contain an indicator for indicating the presence of corrosion.

7. The method of claim 6, wherein the indicator changes color in the presence of the alkaline condition or fluoresces in the presence of the corrosion.

8. The method of claim 5, wherein the coating vehicle is selected from the group consisting of clear paint, clear varnish, opaque varnish, and opaque paint.

9. The method of claim 5, wherein the microcapsules containing the corrosion inhibitor are first microcapsules, and further comprising releasing an indicator from second microcapsules in response to these second microcapsules being exposed to the alkaline condition.

10. A method, comprising:

applying a coating to a metal-containing surface;

wherein the coating comprises first microcapsules dispersed in a coating vehicle, each first microcapsule comprising a first encapsulant contained in a first shell;

wherein the first shells break down in an alkaline condition that naturally occurs in the presence of corrosion to release the first encapsulant; and

5

wherein the first encapsulant includes at least one material selected from the group consisting of corrosion indicators, corrosion inhibitors, and film-forming compounds.

11. The method of claim **10** further comprises:

wherein the coating further comprises second microcapsules dispersed in the coating vehicle, each second microcapsule comprising a second encapsulant contained in a second shell;

wherein the second shells break down in an alkaline condition that naturally occurs in the presence of corrosion to release the second encapsulant; and

wherein the second encapsulant is of a different composition than the first encapsulant and includes at least one material selected from the group consisting of corrosion indicators, corrosion inhibitors, and film-forming compounds.

12. The method of claim **10** further comprises:

wherein the coating further comprises second microcapsules dispersed in the coating vehicle, each second microcapsule comprising a second encapsulant contained in a second shell;

wherein the second shells break down in an alkaline condition that naturally occurs in the presence of corrosion to release the second encapsulant; and

6

wherein the first encapsulant contains a corrosion indicator and the second encapsulant contains a corrosion inhibitor or a film-forming compound or both.

13. The method of claim **10**, wherein the metal-containing surface is a bare metal surface or a coated metal surface.

14. The method of claim **10**, wherein the metal-containing surface comprises a metal or metal alloy.

15. The method of claim **10** further comprises:

wherein the coating further comprises second microcapsules dispersed in the coating vehicle, each second microcapsule comprising a second encapsulant contained in a second shell;

wherein the second shells break down in an alkaline condition that naturally occurs in the presence of corrosion to release the second encapsulant; and

wherein the first and second shells are the same material.

16. The method of claim **10**, wherein the shells break down at pH values of about 8 or above.

17. The method of claim **10**, further comprises indicating a presence corrosion or protecting against corrosion or both.

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